Table 2. Bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses

| $\mathrm{N}(1)-\mathrm{N}(2)$ | $1.409(8)$ | $\mathrm{N}(1)-\mathrm{C}(12)$ | $1.328(9)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N}(2)-\mathrm{C}(3)$ | $1.380(9)$ | $\mathrm{N}(2)-\mathrm{C}(15)$ | $1.411(9)$ |
| $\mathrm{C}(3)-\mathrm{C}(13)$ | $1.417(10)$ | $\mathrm{C}(3)-\mathrm{O}(14)$ | $1.265(9)$ |
| $\mathrm{C}(4)-\mathrm{N}(5)$ | $1.336(9)$ | $\mathrm{C}(4)-\mathrm{C}(13)$ | $1.363(10)$ |
| $\mathrm{N}(5)-\mathrm{C}(11)$ | $1.399(9)$ | $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.385(10)$ |
| $\mathrm{C}(6)-\mathrm{C}(11)$ | $1.388(10)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.375(11)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.376(11)$ | $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.409(10)$ |
| $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.409(10)$ | $\mathrm{C}(10)-\mathrm{C}(12)$ | $1.437(10)$ |
| $\mathrm{C}(12)-\mathrm{C}(13)$ | $1.441(10)$ | $\mathrm{C}(15)-\mathrm{S}(16)$ | $1.705(7)$ |
| $\mathrm{C}(15)-\mathrm{C}(19)$ | $1.386(10)$ | $\mathrm{S}(16)-\mathrm{C}(17)$ | $1.724(7)$ |
| $\mathrm{C}(17)-\mathrm{C}(18)$ | $1.336(11)$ | $\mathrm{C}(17)-\mathrm{C}(20)$ | $1.506(11)$ |
| $\mathrm{C}(18)-\mathrm{C}(19)$ | $1.411(11)$ |  |  |
| $\mathrm{N}(2)-\mathrm{N}(1)-\mathrm{C}(12)$ | $103.2(5)$ | $\mathrm{N}(1)-\mathrm{N}(2)-\mathrm{C}(3)$ | $114.5(5)$ |
| $\mathrm{N}(11-\mathrm{N}(2)-\mathrm{C}(15)$ | $119.0(5)$ | $\mathrm{C}(3)-\mathrm{N}(2)-\mathrm{C}(15)$ | $126.4(6)$ |
| $\mathrm{N}(2)-\mathrm{C}(3)-\mathrm{C}(13)$ | $103.6(6)$ | $\mathrm{N}(2)-\mathrm{C}(3)-\mathrm{O}(14)$ | $124.5(6)$ |
| $\mathrm{C}(13)-\mathrm{C}(3)-\mathrm{O}(14)$ | $131.9(7)$ | $\mathrm{N}(5)-\mathrm{C}(4)-\mathrm{C}(13)$ | $118.8(6)$ |
| $\mathrm{C}(4)-\mathrm{N}(5)-\mathrm{C}(11)$ | $123.5(6)$ | $\mathrm{C}(7)-\mathrm{C}(6)-\mathrm{C}(11)$ | $118.9(6)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $120.9(7)$ | $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | $121.0(7)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | $119.8(7)$ | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $118.3(6)$ |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(12)$ | $124.9(6)$ | $\mathrm{C}(11)-\mathrm{C}(10)-\mathrm{C}(12)$ | $116.8(6)$ |
| $\mathrm{N}(5)-\mathrm{C}(11)-\mathrm{C}(6)$ | $118.5(6)$ | $\mathrm{N}(5)-\mathrm{C}(11)-\mathrm{C}(10)$ | $120.4(6)$ |
| $\mathrm{C}(6)-\mathrm{C}(11)-\mathrm{C}(10)$ | $121.1(6)$ | $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(10)$ | $129.0(6)$ |
| $\mathrm{N}(1)-\mathrm{C}(12)-\mathrm{C}(13)$ | $112.3(6)$ | $\mathrm{C}(10)-\mathrm{C}(12)-\mathrm{C}(13)$ | $118.7(6)$ |
| $\mathrm{C}(3)-\mathrm{C}(13)-\mathrm{C}(4)$ | $131.8(7)$ | $\mathrm{C}(3)-\mathrm{C}(13)-\mathrm{C}(12)$ | $106.3(6)$ |
| $\mathrm{C}(4)-\mathrm{C}(13)-\mathrm{C}(12)$ | $121.8(6)$ | $\mathrm{N}(2)-\mathrm{C}(15)-\mathrm{S}(16)$ | $120.5(5)$ |
| $\mathrm{N}(2)-\mathrm{C}(15)-\mathrm{C}(19)$ | $127.8(6)$ | $\mathrm{S}(16)-\mathrm{C}(15)-\mathrm{C}(19)$ | $111.7(5)$ |
| $\mathrm{C}(15)-\mathrm{S}(16)-\mathrm{C}(17)$ | $91.3(3)$ | $\mathrm{S}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $112.0(6)$ |
| $\mathrm{S}(16)-\mathrm{C}(17)-\mathrm{C}(20)$ | $121.5(5)$ | $\mathrm{C}(18)-\mathrm{C}(17)-\mathrm{C}(20)$ | $126.5(7)$ |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | $113.5(7)$ | $\mathrm{C}(15)-\mathrm{C}(19)-\mathrm{C}(18)$ | $111.4(7)$ |

Fig. I. Perspective view of the title compound, with the atomnumbering system.
factors were calculated by $\sum\left[a_{i} \exp \left(-b_{i} \lambda^{-2} \sin ^{2} \theta\right)\right]+c$ ( $i=1,2,3,4$ ) (International Tables for $X$-ray Crystallography, 1974). Calculations were performed on a FACOM M340R computer at Shionogi Research Laboratories. Final atomic coordinates and equivalent isotropic temperature factors are given in Table 1. Bond distances and angles are listed in Table 2.* A perspective view of the molecule with the atomnumbering system, drawn using the program PLUTO (Motherwell \& Clegg, 1978), is presented in Fig. 1.

Related literature. The structure of the title compound has been discussed by Shindo, Takada, Murata, Eigyo \& Matsushita (1989).

* Lists of H -atom coordinates, anisotropic temperature factors of the non-H atoms and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52627 ( 17 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.


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# Structure of 4-Piperidone Derivatives. I. 3-Methyl-2,6-diphenyl-4-piperidone 

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Abstract. \(\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}, M_{r}=265 \cdot 3\), monoclinic, \(C 2 / c, a\)
\(=19.266\) (2), \(\quad b=6.999\) (2), \(c=21.653\) (1) \(\AA, \quad \beta=\)
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* DCB contribution No. 748.

0108-2701/90/061153-03803.00
$94.42(1)^{\circ}, V=2911.07 \AA^{3}, Z=8, D_{x}=1.21 \mathrm{~g} \mathrm{~cm}^{-3}$, $\lambda(\mathrm{CuK} \mathrm{\alpha})=1.5418 \AA, \quad \mu=5.05 \mathrm{~cm}^{-1}, \quad F(000)=$ 1136, $T=295 \mathrm{~K}, \quad R=0.047$ for 1655 unique observed reflections [ $I>3 \sigma(I)$ ]. The 4-piperidone has © 1990 International Union of Crystallography

Table 1. Fractional atomic coordinates and equivalent isotropic temperature factors

| $U_{\mathrm{eq}}=\left(U_{11}+U_{22}+U_{33}\right) / 3$ |  |  |  |  |
| :--- | :---: | ---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\mathrm{eq}}\left(\AA^{2}\right)$ |
|  | $x$ | $0.1452(2)$ | $0.4698(1)$ | $0.047(1)$ |
| $\mathrm{N}(1)$ | $0.6262(1)$ | $0.0457(3)$ | $0.4270(1)$ | $0.049(1)$ |
| $\mathrm{C}(2)$ | $0.6700(1)$ | $0.04521(3)$ | $0.4180(1)$ | $0.060(1)$ |
| $\mathrm{C}(3)$ | $0.6439(1)$ | $-0.16246(3)$ | $0.4805(1)$ | $0.060(1)$ |
| $\mathrm{C}(4)$ | $0.6434(1)$ | $-0.25464(3)$ | $0.5280(1)$ | $0.061(1)$ |
| $\mathrm{C}(5)$ | $0.6073(1)$ | $-0.1444(3)$ |  |  |
| $\mathrm{C}(6)$ | $0.6331(1)$ | $0.0634(3)$ | $0.5323(1)$ | $0.049(1)$ |
| $\mathrm{C}(7)$ | $0.6678(1)$ | $0.1474(3)$ | $0.3652(1)$ | $0.047(1)$ |
| $\mathrm{C}(8)$ | $0.6058(1)$ | $0.1776(4)$ | $0.3301(1)$ | $0.061(1)$ |
| $\mathrm{C}(9)$ | $0.6053(1)$ | $0.2647(4)$ | $0.2718(1)$ | $0.069(1)$ |
| $\mathrm{C}(10)$ | $0.6665(2)$ | $0.3209(4)$ | $0.2493(1)$ | $0.069(1)$ |
| $\mathrm{C}(11)$ | $0.7279(1)$ | $0.2930(4)$ | $0.2842(1)$ | $0.070(1)$ |
| $\mathrm{C}(12)$ | $0.7286(1)$ | $0.2052(4)$ | $0.3418(1)$ | $0.056(1)$ |
| $\mathrm{C}(13)$ | $0.6852(1)$ | $-0.2755(4)$ | $0.3737(1)$ | $0.088(1)$ |
| $\mathrm{O}(14)$ | $0.6701(1)$ | $-0.4101(2)$ | $0.4923(1)$ | $0.074(1)$ |
| $\mathrm{C}(15)$ | $0.5927(1)$ | $0.1739(3)$ | $0.5774(1)$ | $0.047(1)$ |
| $\mathrm{C}(16)$ | $0.5959(1)$ | $0.1174(4)$ | $0.6389(1)$ | $0.063(1)$ |
| $\mathrm{C}(17)$ | $0.5590(1)$ | $0.2125(4)$ | $0.6817(1)$ | $0.068(1)$ |
| $\mathrm{C}(18)$ | $0.5189(1)$ | $0.3678(3)$ | $0.6642(1)$ | $0.063(1)$ |
| $\mathrm{C}(19)$ | $0.5161(1)$ | $0.4284(4)$ | $0.6040(1)$ | $0.061(1)$ |
| $\mathrm{C}(20)$ | $0.5527(1)$ | $0.3330(3)$ | $0.5608(1)$ | $0.050(1)$ |

Table 2. Bond lengths $(\AA)$, bond angles $\left({ }^{\circ}\right)$ and torsion angles $\left({ }^{\circ}\right)$

| $\mathrm{N}(1)-\mathrm{C}(2)$ | 1.475 (3) | $\mathrm{C}(12)-\mathrm{C}(7)$ | 1-373 (3) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.546 (3) | $\mathrm{C}(3)-\mathrm{C}(13)$ | 1.517 (4) |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.501 (3) | $\mathrm{C}(4)-\mathrm{O}(14)$ | 1-222 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | 1.499 (3) | $\mathrm{C}(6)-\mathrm{N}(1)$ | 1.466 (3) |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.538 (3) | $\mathrm{C}(6)-\mathrm{C}(15)$ | 1.508 (3) |
| $\mathrm{C}(2)-\mathrm{C}(7)$ | 1.513 (3) | $\mathrm{C}(15)-\mathrm{C}(16)$ | 1.386 (3) |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | 1.382 (3) | $\mathrm{C}(16)-\mathrm{C}(17)$ | $1 \cdot 382$ (3) |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | 1.401 (3) | $\mathrm{C}(17)-\mathrm{C}(18)$ | $1 \cdot 370$ (3) |
| $\mathrm{C}(9)-\mathrm{C}(10)$ | 1.368 (4) | $\mathrm{C}(18)-\mathrm{C}(19)$ | $1 \cdot 368$ (3) |
| $\mathrm{C}(10)-\mathrm{C}(11)$ | 1.368 (4) | C(19)-C(20) | $1 \cdot 386$ (3) |
| $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.390 (3) | $\mathrm{C}(20)-\mathrm{C}(15)$ | $1 \cdot 386$ (3) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 109.1 (2) | $\mathrm{C}(12)-\mathrm{C}(7)-\mathrm{C}(2)$ | 119.9 (2) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 108.5 (2) | $\mathrm{C}(7)-\mathrm{C}(2)-\mathrm{C}(3)$ | 110.1 (2) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $115 \cdot 6$ (2) | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(13)$ | 113.0 (2) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $111 \cdot 3$ (2) | $\mathrm{C}(13)-\mathrm{C}(3)-\mathrm{C}(4)$ | 112.6 (2) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{N}(1)$ | 107.9 (2) | $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{O}(14)$ | 122.7 (2) |
| $\mathrm{C}(6)-\mathrm{N}(1)-\mathrm{C}(2)$ | 112.3 (2) | $\mathrm{O}(14)-\mathrm{C}(4)-\mathrm{C}(5)$ | 121.7 (2) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{C}(7)$ | $110 \cdot 8$ (2) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(15)$ | 110.1 (2) |
| $\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{C}(8)$ | 121.5 (2) | $\mathrm{C}(6)-\mathrm{C}(15)-\mathrm{C}(16)$ | 119.3 (2) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | $120 \cdot 5$ (2) | $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | 121.3 (2) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 120.0 (2) | $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $120 \cdot 3$ (2) |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $119 \cdot 7$ (2) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 119.3 (2) |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | $120 \cdot 5$ (2) | $\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(20)$ | $120 \cdot 6$ (2) |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(7)$ | $120 \cdot 8$ (2) | $\mathrm{C}(20)-\mathrm{C}(15)-\mathrm{C}(6)$ | $123 \cdot 3$ (2) |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(15)$ | 112.5 (2) |  |  |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | -55.3 (2) | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{N}(1)$ | 53.3 (2) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 50.5 (2) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{N}(1)-\mathrm{C}(2)$ | -62.7 (2) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | -50.4 (3) | $\mathrm{C}(6)-\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 65.0 (2) |

a slightly distorted chair conformation, the mean torsion angle being $56 \cdot 2^{\circ}$; the puckering is enhanced in the area of $\mathrm{N}(1)$ and decreased in the area of $\mathrm{C}(4)$. The phenyl rings are planar and all molecular dimensions and van der Waals interactions are normal.

Experimental. Crystals were grown from ethanol at room temperature. Data were collected for a colourless transparent crystal $(0.20 \times 0.20 \times$ 0.30 mm ) with an Enraf-Nonius CAD-4 diffractom-
eter using Ni -filtered $\mathrm{Cu} \mathrm{K} \alpha$ radiation. Unit-cell parameters were derived from a least-squares analysis of 25 reflections with $25 \leq 2 \theta \leq 35^{\circ}$. Intensity data were collected with the $\omega-2 \theta$ scan technique, 2003 reflections ( $h-21 \rightarrow 21, k 0 \rightarrow 7, l 0 \rightarrow 24$ ) up to $\theta$ $=60^{\circ}$ were measured, of which 1655 had intensities greater than $3 \sigma(I)$. During data collection three standard reflections, monitored after every 2 h of X-ray exposure, indicated no decay over the full 25 h period. The intensity data were corrected for Lp and for absorption ( $T_{\min }=0.9570$ and $T_{\max }=0.9984$ ). From the observed systematic absences and the statistical test for a centre of symmetry the space group is $C 2 / c$. The structure was solved by direct methods using SHELXS86 (Sheldrick, 1986) and refined on $F$ by weighted full-matrix least squares on a MicroVAX II computer with SHELX76 (Sheldrick, 1976). The hydrogen atoms were located from a difference Fourier map except $\mathrm{C} 2 \mathrm{H}, \mathrm{C} 5 \mathrm{H}$ and C 8 H which were fixed geometrically. All hydrogen atoms were allowed to refine isotropically in final cycles. Final maximum $\Delta / \sigma=0.06$. Maximum and minimum heights in final difference Fourier synthesis $=0.16$ and $-0.18 \mathrm{e} \AA^{-3}$ respectively. Refinement with weights given by $w=1 \cdot 0000 /\left[\sigma^{2}(F)+0 \cdot 000134\left(F_{o}\right)^{2}\right]$ converged at $R=4.7 \%$ ( $w R=5 \cdot 0 \%$ ). Atomic scattering factors were those of SHELX. Final atomic parameters are listed in Table 1 and bond lengths


Fig. 1. A view oŕ the molecule with atom numbering.


Fig. 2. Packing diagram.
and angles in Table 2.* A PLUTO (Motherwell \& Clegg, 1978) drawing of the molecule showing the molecular geometry is presented in Fig. 1, molecular packing in the unit cell in Fig. 2.

Related literature. The 4-piperidone has a slightly distorted chair conformation; puckering is enhanced in the area of $N(1)$ and decreased in the area of $C(4)$. A similar conformational feature is also observed in the 4 -piperidone rings of 3,5 -dimethyl-2,6-di( $p$ -methoxyphenyl)-4-piperidone (Sekar, Parthasarathy \& Radhakrishnan, 1990) and 1,1'-di(4-pyridyl)-

[^0]2,2',6,6'-bi(4-piperidone) dihydrochloride dihydrate (Cheer, Cosgrove \& Vittimberga, 1984).

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# Dendrobine Support Studies. The Structure of a Novel 3-Azatricyclo[6.2.1.0 $\left.{ }^{4,11}\right]$ undecane Derivative 

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#### Abstract

S^{*}, 4 R^{*}, 8 S^{*}, 11 R^{*}\right)\)-3-Diphenylmethyl11 -methyl-3-azatricyclo[6.2.1.0 ${ }^{4,11}$ ]undec-5-en-2-one, $\mathrm{C}_{24} \mathrm{H}_{25} \mathrm{NO}, \quad M_{r}=343 \cdot 47$, monoclinic, $P 2_{1}, \quad a=$ 10.248 (3), $\quad b=8.859$ (2), $\quad c=10.344$ (2) $\AA, \quad \beta=$ $99.816(14)^{\circ}, \quad V=925.4(3) \AA^{3}, \quad Z=2, \quad D_{x}=$ $1.23 \mathrm{~g} \mathrm{~cm}^{-1}(163 \mathrm{~K}), \quad \lambda($ Мо $K \alpha)=0.7107 \AA, \quad \mu=$ $0.6927 \mathrm{~cm}^{-1}, \quad F(000)=368, \quad T=163 \mathrm{~K}, \quad R=0.0423$ for 2569 reflections. The compound is spontaneously resolved upon crystallization. The N atom appears to be $s p^{2}$ hybridized [ N is 0.0901 (13) $\AA$ from plane through three ligand atoms] and in conjugation with the carbonyl group [short N-C bond 1.351 (2) $\AA$ ]. The tricyclic ring system is concave. Ring strain appears to affect $\mathrm{C}-\mathrm{C}$ bond lengths of the central atom of the 3 -ring system. The average $\mathrm{C}-\mathrm{C}$ bond length for this atom to other ring atoms is 1.551 (2) while the average for all other $s p^{3} \mathrm{C}-s p^{3} \mathrm{C}$ bonds is 1.530 (2) $\AA$.

Experimental. (1) was synthesized by an intramolecular Diels-Alder reaction of the corresponding tri-

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enamide. Details of the synthetic procedure have been published elsewhere (Martin \& Li, 1989). Colorless crystals of (1) (m.p. 423-424 K) were

(1)
obtained by slow evaporation from ethyl acetatehexanes (1:19) from which (1) resolves spontaneously. The data crystal was cut from a large block and had approximate dimensions $0.35 \times 0.37 \times 0.45 \mathrm{~mm}$. The data were collected on a Syntex $P 2_{1}$ diffractometer, with a graphite monochromator, and equipped with a Syntex LT-1 low temperature delivery system ( 163 K ). Lattice parameters were obtained from the least-squares refinement of 45 reflections with $24 \cdot 2<$ $2 \theta<31 \cdot 2^{\circ}$. Data were collected using the $\omega$-scan
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[^0]:    * Lists of structure factors, anisotropic thermal parameters, H -atom parameters, bond lengths and angles involving H atoms and least-squares-planes data, have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52599 ( 16 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

